Summer Reading Assignment for Honors Physics.

DIRECTIONS:

1. Scroll below to view the summer reading packet.
2. Read the entire assignment.
3. Practice the example problems and exercises within the reading assignment.
4. Answer the following questions at the end of the packet:
   #1,3,9,17,21,23,25,35,37
5. This assignment will be collected on the first day of class. Show all work!!
The goal of physics is to gain a deeper understanding of the world in which we live. For example, the laws of physics allow us to predict the behavior of everything from rockets sent to the Moon, to integrated chips in computers, to lasers used to perform eye surgery. In short, everything in nature—from atoms and subatomic particles to solar systems and galaxies—obeys the laws of physics.

As we begin our study of physics, it is useful to consider a range of issues that underlie everything to follow. One of the most fundamental of these is the system of units we use when we measure such things as the mass of an object, its length, and the time between two events. Other equally important issues include methods for handling numerical calculations and basic conventions of mathematical
notation. By the end of the chapter we will have developed a common "language" of physics that will be used throughout this book and probably in any science that you study.

1-1 Physics and the Laws of Nature

Physics is the study of the fundamental laws of nature, which, simply put, are the laws that underlie all physical phenomena in the universe. Remarkably, we have found that these laws can be expressed in terms of mathematical equations. As a result, it is possible to make precise, quantitative comparisons between the predictions of theory—derived from the mathematical form of the laws—and the observations of experiments. Physics, then, is a science rooted equally firmly in theory and experiment.

What makes physics particularly fascinating is the fact that it relates to everything in the universe. There is a great beauty in the vision that physics brings to our view of the universe; namely, that all the complexity and variety that we see in the world around us, and in the universe as a whole, are manifestations of a few fundamental laws and principles. That we can discover and apply these basic laws of nature is both astounding and exhilarating.

For those not familiar with the subject, physics may seem to be little more than a confusing mass of formulas. Sometimes, in fact, these formulas can be the trees that block the view of the forest. For a physicist, however, the many formulas of physics are simply different ways of expressing a few fundamental ideas. It is the forest—the basic laws and principles of physical phenomena in nature—that is the focus of this text.

1-2 Units of length, Mass, and Time

To make quantitative comparisons between the laws of physics and our experience of the natural world, certain basic physical quantities must be measured. The most common of these quantities are length (L), mass (M), and time (T). In fact, in the next several chapters, these are the only quantities that arise. Later in the text, additional quantities, such as temperature and electric current, will be introduced as needed.

We begin by defining the units in which each of these quantities is measured. Once the units are defined, the values obtained in specific measurements can be expressed as multiples of them. For example, our unit of length is the meter (m). It follows, then, that a person who is 1.94 m tall has a height 1.94 times this unit of length. Similar comments apply to the unit of mass, the kilogram, and the unit of time, the second.

The detailed system of units used in this book was established in 1960 at the Eleventh General Conference of Weights and Measures in Paris, France, and goes by the name Système International, or SI for short. Thus, when we refer to SI units, we mean units of meters (m), kilograms (kg), and seconds (s). Taking the first letter from each of these units leads to an alternate name that is often used—the mks system.

In the remainder of this section we define each of the SI units.

Length

Early units of length were often associated with the human body. For example, the Egyptians defined the cubit to be the distance from the elbow to the tip of the middle finger. Similarly, the foot was at one time defined to be the length of the royal foot of King Louis XIV. As colorful as these units may be, they are not particularly reproducible—at least not to great precision.

In 1793, the French Academy of Sciences, seeking a more objective and reproducible standard, decided to define a unit of length equal to one ten-millionth the
distance from the North Pole to the equator. This new unit was named the metre (from the Greek metron for “measure”). The preferred spelling in the United States is meter. This definition was widely accepted, and in 1799 a “standard” meter was produced. It consisted of a platinum–iridium alloy rod with two marks on it one meter apart.

Since 1983, however, we have used an even more precise definition of the meter, based on the speed of light in a vacuum. In particular:

One meter is defined to be the distance traveled by light in a vacuum in $1/299,792,458$ of a second.

No matter how its definition is refined, however, a meter is still about 3.28 feet, which is roughly 10 percent longer than a yard. A list of typical lengths is given in Table 1-1.

| Distance from Earth to the nearest large galaxy (the Andromeda galaxy, M31) | $2 \times 10^{22}$ m |
| Diameter of our galaxy (the Milky Way) | $8 \times 10^{20}$ m |
| Distance from Earth to the nearest star (other than the sun) | $4 \times 10^{16}$ m |
| One light year | $9.46 \times 10^{15}$ m |
| Average radius of Pluto’s orbit | $6 \times 10^{13}$ m |
| Distance from Earth to the Sun | $1.5 \times 10^{11}$ m |
| Radius of Earth | $6.37 \times 10^{6}$ m |
| Length of football field | $10^{2}$ m |
| Height of a person | 2 m |
| Diameter of a CD | 0.12 m |
| Diameter of the aorta | 0.018 m |
| Diameter of a period in a sentence | $5 \times 10^{-4}$ m |
| Diameter of a red blood cell | $8 \times 10^{-6}$ m |
| Diameter of the hydrogen atom | $10^{-10}$ m |
| Diameter of a proton | $2 \times 10^{-13}$ m |

The size of these viruses, seen here attacking a bacterial cell, is about $10^{-7}$ m.

The diameter of this typical galaxy is about $10^{21}$ m. (How many viruses would it take to span the galaxy?)
## Mass

In SI units, mass is measured in kilograms. Unlike the meter, the kilogram is not based on any natural physical quantity. By convention, the kilogram has been defined as follows:

The kilogram, by definition, is the mass of a particular platinum-iridium alloy cylinder at the International Bureau of Weights and Standards in Sèvres, France.

To put the kilogram in everyday terms, a quart of milk has a mass slightly less than 1 kilogram. Additional masses, in kilograms, are given in Table 1-2.

Note that we do not define the kilogram to be the weight of the platinum-iridium cylinder. In fact, weight and mass are quite different quantities, even though they are often confused in everyday language. Mass is an intrinsic, unchanging property of an object. Weight, in contrast, is a measure of the gravitational force acting on an object, which can vary depending on the object’s location. For example, if you are fortunate enough to travel to Mars someday, you will find that your weight is less than on Earth, though your mass is unchanged. The force of gravity will be discussed in detail in Chapter 12.

### Time

Nature has provided us with a fairly accurate timepiece in the revolving Earth. In fact, prior to 1956, the mean solar day was defined to consist of 24 hours, with 60 minutes per hour, and 60 seconds per minute, for a total of \((24)(60)(60) = 86,400\) seconds. Even the rotation of the Earth is not completely regular, however.

Today, the most accurate timekeepers known are “atomic clocks,” which are based on characteristic frequencies of radiation emitted by certain atoms. These clocks have typical accuracies of about 1 second in 300,000 years. The atomic clock used for defining the second operates with cesium-133 atoms. In particular, the second is defined as follows:

One second is defined to be the time it takes for radiation from a cesium-133 atom to complete 9,192,631,770 cycles of oscillation.

A range of characteristic time intervals is given in Table 1-3.

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**TABLE 1-2 Typical Masses**

<table>
<thead>
<tr>
<th>Object</th>
<th>Mass (kg)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Galaxy (Milky Way)</td>
<td>(4 \times 10^{41})</td>
</tr>
<tr>
<td>Sun</td>
<td>(2 \times 10^{30})</td>
</tr>
<tr>
<td>Earth</td>
<td>(5.97 \times 10^{24})</td>
</tr>
<tr>
<td>Space Shuttle</td>
<td>(2 \times 10^{6})</td>
</tr>
<tr>
<td>Elephant</td>
<td>5400 kg</td>
</tr>
<tr>
<td>Automobile</td>
<td>1200 kg</td>
</tr>
<tr>
<td>Human</td>
<td>70 kg</td>
</tr>
<tr>
<td>Baseball</td>
<td>15 kg</td>
</tr>
<tr>
<td>Honeybee</td>
<td>1.5 \times 10^{-4} kg</td>
</tr>
<tr>
<td>Red blood cell</td>
<td>10^{-13} kg</td>
</tr>
<tr>
<td>Bacterium</td>
<td>10^{-15} kg</td>
</tr>
<tr>
<td>Hydrogen atom</td>
<td>1.67 \times 10^{-27} kg</td>
</tr>
<tr>
<td>Electron</td>
<td>9.11 \times 10^{-31} kg</td>
</tr>
</tbody>
</table>

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▲ The standard kilogram, a cylinder of platinum and iridium 0.039 m in height and diameter, is kept under carefully controlled conditions in Sèvres, France. Exact replicas are maintained in other laboratories around the world.

▲ This atomic clock, which keeps time on the basis of radiation from cesium atoms, is accurate to about three millionths of a second per year. (How long would it take for it to gain or lose an hour?)
TABLE 1-3 Typical Times

<table>
<thead>
<tr>
<th>Event</th>
<th>Time</th>
</tr>
</thead>
<tbody>
<tr>
<td>Age of the universe</td>
<td>$5 \times 10^{27}$ s</td>
</tr>
<tr>
<td>Age of the Earth</td>
<td>$1.3 \times 10^{17}$ s</td>
</tr>
<tr>
<td>Existence of human species</td>
<td>$6 \times 10^{37}$ s</td>
</tr>
<tr>
<td>Human lifetime</td>
<td>$2 \times 10^9$ s</td>
</tr>
<tr>
<td>One year</td>
<td>$3 \times 10^7$ s</td>
</tr>
<tr>
<td>One day</td>
<td>$8.6 \times 10^4$ s</td>
</tr>
<tr>
<td>Time between heartbeats</td>
<td>0.8 s</td>
</tr>
<tr>
<td>Human reaction time</td>
<td>0.1 s</td>
</tr>
<tr>
<td>One cycle of a high-pitched sound wave</td>
<td>$5 \times 10^{-3}$ s</td>
</tr>
<tr>
<td>One cycle of an AM radio wave</td>
<td>$10^{-6}$ s</td>
</tr>
<tr>
<td>One cycle of a visible light wave</td>
<td>$2 \times 10^{-15}$ s</td>
</tr>
</tbody>
</table>

Other Systems of Units and Standard Prefixes

Although SI units are used throughout most of this book and are used almost exclusively in scientific research and industry, we will occasionally refer to other systems that you may encounter from time to time.

For example, a system of units similar to the mks system, though comprised of smaller units, is the cgs system, which stands for centimeter (cm), gram (g), and second (s). In addition, the British engineering system is often encountered in everyday usage in the United States. Its basic units are the slug for mass, the foot (ft) for length, and the second (s) for time.

Finally, multiples of the basic units are common no matter which system is used. Standard prefixes are used to designate common multiples in powers of ten. For example, the prefix kilo means one thousand, or, equivalently, $10^3$. Thus, 1 kilogram is $10^3$ grams, and 1 kilometer is $10^3$ meters. Similarly, milli is the prefix for one thousandth, or $10^{-3}$. Thus, a millimeter is $10^{-3}$ meter, and so on. The most common prefixes are listed in Table 1–4.

**TABLE 1–4 Common Prefixes**

<table>
<thead>
<tr>
<th>Power</th>
<th>Prefix</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>$10^{15}$</td>
<td>peta</td>
<td>P</td>
</tr>
<tr>
<td>$10^{12}$</td>
<td>tera</td>
<td>T</td>
</tr>
<tr>
<td>$10^9$</td>
<td>giga</td>
<td>G</td>
</tr>
<tr>
<td>$10^6$</td>
<td>mega</td>
<td>M</td>
</tr>
<tr>
<td>$10^3$</td>
<td>kilo</td>
<td>k</td>
</tr>
<tr>
<td>$10^2$</td>
<td>hecto</td>
<td>h</td>
</tr>
<tr>
<td>$10^1$</td>
<td>deca</td>
<td>da</td>
</tr>
<tr>
<td>$10^{-1}$</td>
<td>deci</td>
<td>d</td>
</tr>
<tr>
<td>$10^{-2}$</td>
<td>centi</td>
<td>c</td>
</tr>
<tr>
<td>$10^{-3}$</td>
<td>milli</td>
<td>m</td>
</tr>
<tr>
<td>$10^{-6}$</td>
<td>micro</td>
<td>μ</td>
</tr>
<tr>
<td>$10^{-9}$</td>
<td>nano</td>
<td>n</td>
</tr>
<tr>
<td>$10^{-12}$</td>
<td>pico</td>
<td>p</td>
</tr>
<tr>
<td>$10^{-15}$</td>
<td>femto</td>
<td>f</td>
</tr>
</tbody>
</table>

**EXERCISE 1–1**

(a) A minivan sells for 33,200 dollars. Express the price of the minivan in kilodollars and megadollars.

(b) A typical E. coli bacterium is about 5 micrometers (or microns) in length. Give this length in millimeters and kilometers.

**Solution**

(a) 33.2 kilodollars, 0.0332 megadollars.

(b) 0.005 mm, 0.000000005 km.

**1–3 Dimensional Analysis**

In physics, when we speak of the dimension of a physical quantity, we refer to the type of quantity in question, regardless of the units used in the measurement. For example, a distance measured in cubits and another distance measured in light-years both have the same dimension—length. The same is true of compound units such as velocity, which has the dimensions of length per unit time (length/time).

A velocity measured in miles per hour has the same dimensions—length/time—as one measured in inches per century.

Now, any valid formula in physics must be dimensionally consistent; that is, each term in the equation must have the same dimensions. It simply doesn’t make sense to add a distance to a time, for example, any more than it makes sense to add apples and oranges. They are different things.
To check the dimensional consistency of an equation, it is convenient to introduce a special notation for the dimension of a quantity. We will use square brackets, \([\cdot]\), for this purpose. Thus, if \(x\) represents a distance, which has dimensions of length \([L]\), we write this as \(x = [L]\). Similarly, a velocity, \(v\), has dimensions of length per time \([LT]\); thus we write \(v = [L]/[T]\) to indicate its dimensions. Acceleration, \(a\), which is the change in velocity per time, has the dimensions \(a = ([L]/[T])/[T] = [L]/[T^2]\). The dimensions of some common physical quantities are summarized in Table 1-5.

Let's use this notation to check the dimensional consistency of a simple equation. Consider the following formula:

\[ x = x_0 + vt \]

In this equation, \(x\) and \(x_0\) represent distances, \(v\) is a velocity, and \(t\) is time. Writing out the dimensions of each term, we have

\[ [L] = [L] + [L][T][T] \]

It might seem at first that the last term has different dimensions than the other two. However, dimensions obey the same rules of algebra as other quantities. Thus the dimensions of time cancel in the last term:

\[ [L] = [L] + [L][T][T] = [L] + [L] \]

As a result, we see that each term in this formula has the same dimensions. This type of calculation with dimensions is referred to as dimensional analysis.

**EXERCISE 1-2**

Show that \(x = x_0 + v_0 t + \frac{1}{2}at^2\) is dimensionally consistent. The quantities \(x\) and \(x_0\) are distances, \(v_0\) is a velocity, and \(a\) is an acceleration.

**Solution**

Using the dimensions given in Table 1-5 we have

\[ [L] = [L] + [L][T][T] + [L][T^2][T^2] = [L] + [L] + [L] \]

Note that \(\frac{1}{2}\) is ignored in this analysis because it has no dimensions.

Later in this text you will derive your own formulas from time to time. As you do so, it is helpful to check dimensional consistency at each step of the derivation. If at any time the dimensions don’t agree, you will know that a mistake has been made, and you can go back and look for it. If the dimensions check, however, it’s not a guarantee the formula is correct—after all, dimensionless factors, like 1/2 or 2, don’t show up in a dimensional check.

**1-4 Significant Figures**

When a mass, a length, or a time is measured in a scientific experiment, the result is known only to within a certain accuracy. The inaccuracy or uncertainty can be caused by a number of factors, ranging from limitations of the measuring device itself to limitations associated with the senses and the skill of the person performing the experiment. In any case, the fact that observed values of experimental quantities have inherent uncertainties should always be kept in mind when performing calculations with those values.
Suppose, for example, that you want to determine the walking speed of your pet tortoise. To do so, you measure the time, $t$, it takes for the tortoise to walk a distance, $d$, and then you calculate the quotient, $d/t$. When you measure the distance with a ruler, which has one tick mark per millimeter, you find that $d = 21.2$ cm, with the precise value of the digit in the second decimal place uncertain. Defining the number of significant figures in a physical quantity to be equal to the number of digits in it that are known with certainty, we say that $d$ is known to three significant figures.

Similarly, you measure the time with an old pocket watch, and as best you can determine it, $t = 8.5$ s, with the second decimal place uncertain. Note that $t$ is known to only two significant figures. If we were to make this measurement with a digital watch, with a readout giving the time to $1/100$ of a second, the accuracy of the result would still be limited by the finite reaction time of the experimenter. The reaction time would have to be predetermined in a separate experiment (See Problem 67 in Chapter 2 for a simple way to determine your reaction time.)

Returning to the problem at hand, we would now like to calculate the speed of the tortoise. Using the above values for $d$ and $t$ and a calculator with eight digits in its display, we find $(21.2 \text{ cm})/(8.5 \text{ s}) = 2.4941176 \text{ cm/s}$. Clearly, such an accurate value for the speed is unjustified, considering the limitations of our measurements. After all, we can't expect to measure quantities to two and three significant figures and from them obtain results with eight significant figures. In general, the number of significant figures that result when we multiply or divide physical quantities is given by the following rule of thumb:

The number of significant figures after multiplication or division is equal to the number of significant figures in the least accurately known quantity.

In our speed calculation, for example, we know the distance to three significant figures, but the time to only two significant figures. As a result, the speed should be given with just two significant figures, $d/t = (21.2 \text{ cm})/(8.5 \text{ s}) = 2.5 \text{ cm/s}$. Note that we didn't just keep the first two digits in 2.4941176 cm/s and drop the rest. Instead, we "rounded up"; that is, because the first digit to be dropped (9 in this case) is greater than or equal to 5, we increase the previous digit (4 in this case) by 1. Thus, 2.5 cm/s is our best estimate for the tortoise's speed.

**EXAMPLE 1-1 It's the Tortoise by a Hare**

A tortoise races a rabbit by walking with a constant speed of 2.51 cm/s for 12.23 s. How much distance does the tortoise cover?

**Picture the Problem**

The race between the rabbit and the tortoise is shown in our sketch. The rabbit pauses to eat a carrot while the tortoise walks with a constant speed.

**Strategy**

The distance covered by the tortoise is the speed of the tortoise multiplied by the time during which it walks.

**Solution**

1. Multiply the speed by the time to find the distance $d$:

$$d = (\text{speed})(\text{time})$$

$$= (2.51 \text{ cm/s})(12.23 \text{ s}) = 30.7 \text{ cm}$$
Insight

Notice that if we simply multiply 2.51 cm/s by 12.23 s, we obtain 30.6973 cm. We don’t give all of these digits in our answer, however. In particular, because the quantity that is known with the least accuracy (the speed) has only three significant figures, we give a result with three significant figures. Note, in addition, that the third digit in our answer has been rounded up from 6 to 7.

Practice Problem

How long does it take for the tortoise to walk 17 cm?  [Answer: \( t = \frac{(17 \text{ cm})}{(2.51 \text{ cm/s})} = 6.8 \text{ s} \)]

Some related homework problems: Problem 11, Problem 15

Note that the distance of 17 cm in the Practice Problem has only two significant figures because we don’t know the digits to the right of the decimal place. If the distance were given as 17.0 cm, on the other hand, it would have three significant figures.

When physical quantities are added or subtracted, we use a slightly different rule of thumb. In this case, the rule involves the number of decimal places in each of the terms:

The number of decimal places after addition or subtraction is equal to the smallest number of decimal places in any of the individual terms.

Thus, if you make a time measurement of 16.74 s, and then a subsequent time measurement of 5.1 s, the total time of the two measurements should be given as 21.8 s, rather than 21.84 s.

EXERCISE 1-3

You and a friend pick some raspberries. Your flat weighs 12.7 lb, and your friend’s weighs 7.25 lb. What is the combined weight of the raspberries?

Solution

Just adding the two numbers gives 19.95 lb. According to our rule of thumb, however, the final result must have only a single decimal place (corresponding to the term with the smallest number of decimal places). Rounding off to one place, then, gives 20.0 lb as the acceptable result.

Scientific Notation

The number of significant figures in a given quantity may be ambiguous due to the presence of zeros at the beginning or end of the number. For example, if a distance is stated to be 2500 m, the two zeros could be significant figures, or they could be zeros that simply show where the decimal point is located. If the two zeros are significant figures, the uncertainty in the distance is roughly a meter; if they are not significant figures, however, the uncertainty is about 100 m.

To remove this type of ambiguity, we can write the distance in scientific notation—that is, as a number of order unity times an appropriate power of ten. Thus, in this example, we would express the distance as 2.5 \( \times \) 10^3 m if there are only two significant figures, or as 2.500 \( \times \) 10^3 m to indicate four significant figures. Likewise, a time given as 0.000 036 s has only two significant figures—the preceding zeros only serve to fix the decimal point. If this quantity were known to three significant figures, we would write it as 3.60 \( \times \) 10^{-3} s to remove any ambiguity. See Appendix A for a more detailed discussion of scientific notation.

EXERCISE 1-4

How many significant figures are there in (a) 21.00, (b) 21, (c) 2.1 \( \times \) 10^{-2}, and (d) 2.10 \( \times \) 10^{-3}?

Solution

(a) 4, (b) 2, (c) 2, (d) 3
Round-Off Error

Finally, even if you perform all your calculations to the same number of significant figures as in the text, you may occasionally obtain an answer that differs in its last digit from that given in the book. In most cases this is not an issue as far as understanding the physics is concerned—usually it is due to round-off error.

Round-off error occurs when numerical results are rounded off at different times during a calculation. To see how this works, let’s consider a simple example. Suppose you are shopping for knickknacks, and you buy one item for $2.21, plus 8 percent sales tax. The total price is $2.3868, or, rounded off to the nearest penny, $2.39. Later, you buy another item for $1.35. With tax this becomes $1.458 or, again to the nearest penny, $1.46. The total expenditure for these two items is $2.39 + $1.46 = $3.85.

Now, let’s do the rounding off in a different way. Suppose you buy both items at the same time for a total before-tax price of $2.21 + $1.35 = $3.56. Adding in the 8 percent tax gives $3.8448, which rounds off to $3.84, one penny different from the previous amount. This same type of discrepancy can occur in physics problems. In general, it’s a good idea to keep one extra digit throughout your calculations whenever possible, rounding off only the final result. But while this practice can help to reduce the likelihood of round-off error, there is no way to avoid it in every situation.

1–5 Converting Units

It is often convenient to convert from one set of units to another. For example, suppose you would like to convert 316 ft to its equivalent in meters. Looking at the conversion factors on the inside front cover of the text, we see that

\[ 1 \text{ m} = 3.281 \text{ ft} \]

Equivalently,

\[ \frac{1 \text{ m}}{3.281 \text{ ft}} = 1 \]

Now, to make the conversion, we simply multiply 316 ft by this expression, which is equivalent to multiplying by 1:

\[ (316 \text{ ft}) \left( \frac{1 \text{ m}}{3.281 \text{ ft}} \right) = 96.3 \text{ m} \]

Note that the conversion factor is written in this particular way, as 1 m divided by 3.281 ft, so that the units of feet cancel out, leaving the final result in the desired units of meters.

Of course, we can just as easily convert from meters to feet if we use the reciprocal of this conversion factor—which is also equal to 1:

\[ 1 = \frac{3.281 \text{ ft}}{1 \text{ m}} \]

For example, a distance of 26.4 m is converted to feet by canceling out the units of meters, as follows:

\[ (26.4 \text{ m}) \left( \frac{3.281 \text{ ft}}{1 \text{ m}} \right) = 86.6 \text{ ft} \]

Thus, we see that converting units is as easy as multiplying by 1—because that’s really what you’re doing.

\[ \text{From this sign, you can calculate factors for converting miles to kilometers and vice versa. (Why do you think the conversion factors seem to vary for different destinations?)} \]
EXAMPLE 1–2 A High-Volume Warehouse

A warehouse is 20.0 yards long, 10.0 yards wide, and 15.0 feet high. What is its volume in SI units?

Picture the Problem
In our sketch we picture the warehouse and indicate the relevant length for each of its dimensions.

Strategy
We begin by converting the length, the width, and the height of the warehouse to meters. Once this is done, the volume in SI units is simply the product of the three dimensions.

Solution
1. Convert the length of the warehouse to meters:
   \[ L = (20.0 \text{ yard}) \left( \frac{3 \text{ ft}}{1 \text{ yard}} \right) \left( \frac{1 \text{ m}}{3.281 \text{ ft}} \right) = 18.3 \text{ m} \]
2. Convert the width to meters:
   \[ W = (10.0 \text{ yard}) \left( \frac{3 \text{ ft}}{1 \text{ yard}} \right) \left( \frac{1 \text{ m}}{3.281 \text{ ft}} \right) = 9.14 \text{ m} \]
3. Convert the height to meters:
   \[ H = (15.0 \text{ ft}) \left( \frac{1 \text{ m}}{3.281 \text{ ft}} \right) = 4.57 \text{ m} \]
4. Calculate the volume of the warehouse:
   \[ V = L \times W \times H = (18.3 \text{ m})(9.14 \text{ m})(4.57 \text{ m}) = 764 \text{ m}^3 \]

Insight
We would say, then, that the warehouse has a volume of 764 cubic meters.

Practice Problem
What is the volume of the warehouse if its length is one-hundredth of a mile? [Answer: \( V = 672 \text{ m}^3 \)]

Some related homework problems: Problem 17, Problem 18

Finally, the same procedure can be applied to conversions involving any number of units. For instance, if you walk at 3.00 mi/h, how fast is that in m/s? In this case we need the following additional conversion factors:

\[ 1 \text{ mi} = 5,280 \text{ ft} \quad 1 \text{ h} = 3,600 \text{ s} \]

With these factors at hand, we carry out the conversion as follows:

\[ (3.00 \text{ mi}/\text{h}) \left( \frac{5,280 \text{ ft}}{1 \text{ mi}} \right) \left( \frac{1 \text{ m}}{3.281 \text{ ft}} \right) \left( \frac{1 \text{ ft}}{3.00 \text{ s}} \right) = 1.34 \text{ m/s} \]

Note that in each conversion factor the numerator is equal to the denominator. In addition, each conversion factor is written in such a way that the unwanted units cancel, leaving just meters per second.

ACTIVE EXAMPLE 1–1 Find the Speed of Blood

Blood in the human aorta can attain speeds of 35.0 cm/s. How fast is this in (a) ft/s and (b) mi/h?

Solution (Test your understanding by performing the calculations indicated in each step.)

Part (a)

1. Convert centimeters to meters
   \[ 1.15 \text{ ft/s} \]

   and then to feet:
Part (b)
1. First, convert centimeters to miles: \(2.17 \times 10^{-4} \text{ mi/s}\)
2. Next, convert seconds to hours: \(0.783 \text{ mi/h}\)

Insight
Of course, the conversions in part (b) can be carried out in a single calculation if desired.

Your Turn
Find the speed of blood in units of km/h.
(Answers to Your Turn problems are given in the back of the book.)

1–6 Order-of-Magnitude Calculations
An order-of-magnitude calculation is a rough, “ballpark” estimate designed to be accurate to within a factor of about 10. One purpose of such a calculation is to give a quick idea of what order of magnitude should be expected from a complete, detailed calculation. If an order-of-magnitude calculation indicates that a distance should be on the order of \(10^4\) m, for example, and your calculator gives an answer on the order of \(10^7\) m, then there is an error somewhere that needs to be resolved.

For example, suppose you would like to estimate the speed of a cliff diver on entering the water. First, the cliff may be 20 or 30 feet high; thus in SI units we would say that the order of magnitude of the cliff’s height is \(10^m\)—certainly not \(1\) m or \(10^2\) m. Next, the diver hits the water something like a second later—certainly not \(0.1\) s later nor \(10\) s later. Thus, a reasonable order-of-magnitude estimate of the diver’s speed is \(10\) m/1 s = \(10\) m/s, or roughly \(20\) mi/h. If you do a detailed calculation and your answer is on the order of \(10^4\) m/s, you probably entered one of your numbers incorrectly.

Another reason for doing an order-of-magnitude calculation is to get a feeling for what size numbers we are talking about in situations where a precise count is not possible. This is illustrated in the following Example.

EXAMPLE 1–3  Estimation: How Many Raindrops in a Storm
A thunderstorm drops a half-inch (\(-0.01\) m) of rain on Washington, D.C., which covers an area of about 70 square miles (\(-10^8\) m\(^2\)). Estimate the number of raindrops that fell during the storm.

Picture the Problem
Our sketch shows an area \(A = 10^8\) m\(^2\) covered to a depth \(d = 0.01\) m by rainwater from the storm. Each drop of rain is approximated by a small sphere with a diameter of 4 mm.

Strategy
To find the number of raindrops, we first calculate the volume of water required to cover \(10^8\) m\(^2\) to a depth of \(0.01\) m. Next, we calculate the volume of an individual drop of rain, recalling that the volume of a sphere of radius \(r\) is \(4\pi r^3/3\). We estimate the diameter of a raindrop to be about 4 mm. Finally, dividing the volume of a drop into the volume of water that fell during the storm gives the number of drops.

continued on following page
1. Calculate the order of magnitude of the volume of water, $V_{\text{water}}$, that fell during the storm:

$$V_{\text{water}} = Ad = (10^8 \text{ m}^2)(0.01 \text{ m}) \approx 10^6 \text{ m}^3$$

2. Calculate the order of magnitude of the volume of a drop of rain, $V_{\text{drop}}$. Note that if the diameter of a drop is 4 mm, its radius is $r = 2 \text{ mm} = 0.002 \text{ m}$:

$$V_{\text{drop}} = \frac{4}{3} \pi r^3 \approx \frac{4}{3} \pi (0.002 \text{ m})^3 \approx 10^{-8} \text{ m}^3$$

3. Divide $V_{\text{drop}}$ into $V_{\text{water}}$ to find the order of magnitude of the number of drops that fell during the storm:

$$\text{number of drops} \approx \frac{V_{\text{water}}}{V_{\text{drop}}} \approx \frac{10^6 \text{ m}^3}{10^{-8} \text{ m}^3} = 10^{14}$$

Insight

The number of raindrops in this one small storm is roughly 100,000 times greater than the current population of Earth.

Practice Problem

If a storm pelts Washington, D.C. with $10^{15}$ raindrops, how many inches of rain fall on the city? [Answer: About 5 inches.]

Some related homework problems: Problem 31, Problem 33

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1–7 Problem Solving in Physics

Physics is a lot like swimming—you have to learn by doing. You could read a book on swimming and memorize every word in it, but when you jump into a pool the first time you are going to have problems. Similarly, you could read this book carefully, memorizing every formula in it, but when you finish, you still haven’t learned physics. To learn physics you have to go beyond passive reading; you have to interact with physics and experience it by doing problems.

In this section we present a general overview of problem solving in physics. The suggestions given below, which apply to problems in all areas of physics, should help to develop a systematic approach.

We should emphasize at the outset that there is no recipe for solving problems in physics—it is a creative activity. In fact, the opportunity to be creative is one of the attractions of physics. The following suggestions, then, are not intended as a rigid set of steps that must be followed like the steps in a computer program. Rather, they provide a general guideline that experienced problem solvers find to be effective.

- **Read the problem carefully** Before you can solve a problem you need to know exactly what information it gives and what it asks you to determine. Some information is given explicitly, as when a problem states that a person has a mass of 70 kg. Other information is implicit; for example, saying that a ball is dropped from rest means that its initial speed is zero. Clearly, a careful reading is the essential first step in problem solving.

- **Sketch the system** This may seem like a step you can skip—but don't. A sketch helps you to acquire a physical feeling for the system. It also provides an opportunity to label those quantities that are known and those that are to be determined. All Examples in this text begin with a sketch of the system, accompanied by a brief description in a section labeled “Picture the Problem.”

- **Visualize the physical process** Try to visualize what is happening in the system as if you were watching it in a movie. Your sketch should help. This step ties in closely with the next step.

- **Strategize** This may be the most difficult, but at the same time the most creative, part of the problem-solving process. From your sketch and visualization, try to
identify the physical processes at work in the system. Then, develop a strategy—a game plan—for solving the problem. All Examples in this book have a “Strategy” spelled out before the solution begins.

• Identify appropriate equations Once a strategy has been developed, find the specific equations that are needed to carry it out.

• Solve the equations Use basic algebra to solve the equations identified in the previous step. Work with symbols such as $x$ or $y$ for the most part, substituting numerical values near the end of the calculations.

• Check your answer Once you have an answer, check to see if it makes sense: (i) Does it have the correct dimensions? (ii) Is the numerical value reasonable?

• Explore limits/special cases Getting the correct answer is nice, but it’s not all there is to physics. You can learn a great deal about physics and about the connection between physics and mathematics by checking various limits of your answer. For example, if you have two masses in your system, $m_1$ and $m_2$, what happens in the special case that $m_1 = 0$ or $m_1 = m_2$? Check to see whether your answer and your physical intuition agree.

The Examples in this text are designed to deepen your understanding of physics and at the same time develop your problem-solving skills. They all have the same basic structure: Problem Statement; Picture the Problem; Strategy; Solution; presenting the flow of ideas and the mathematics side-by-side in a two-column format; Insight; and a Practice Problem related to the one just solved. As you work through the Examples in the chapters to come, notice how the basic problem-solving guidelines outlined above are implemented in a consistent way.

Finally, it is tempting to look for shortcuts when doing a problem—to look for a formula that seems to fit and some numbers to plug into it. It may seem harder to think ahead, to be systematic as you solve the problem, and then to think back over what you have done at the end of the problem. The extra effort is worth it, however, because by doing these things you will develop powerful problem-solving skills that can be applied to unexpected problems you may encounter on exams—and in life in general.

## Chapter Summary

### Remarks and Relevant Equations

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1-4 Significant Figures

- significant figures
- round-off error

The number of digits reliably known, excluding digits that simply indicate the decimal place. For example, 3.45 and 0.0000345 both have three significant figures.

Discrepancies caused by rounding off numbers in intermediate results.

1-5 Converting Units

Multiply by the ratio of two units to convert from one to another. As an example, to convert 3.5 m to feet you multiply by the factor (1 ft/0.3048 m).

1-6 Order-of-Magnitude Calculations

A ballpark estimate designed to be accurate to within the nearest power of ten.

1-7 Problem Solving in Physics

A general approach to problem solving is as follows: read; sketch; visualize; strategize; identify equations; solve; check; explore limits.

Conceptual Questions

(Answers to odd-numbered Conceptual Questions can be found in the back of the book, beginning on page A-21.)

1. Can dimensional analysis determine whether the area of a circle is \( \pi r^2 \) or \( 2\pi r^2 \)? Explain.
2. If a distance \( d \) has units of meters, and a time \( T \) has units of seconds, does the quantity \( T/d \) make sense physically? What about the quantity \( a/d \)? Explain in both cases.
3. Which of the following equations is dimensionally consistent?
   (a) \( x = vt \), (b) \( x = \frac{1}{2}at^2 \), (c) \( t = \frac{(2x/a)}{\sqrt{2}} \).
4. Which of the following equations is dimensionally consistent?
   (a) \( v = at \), (b) \( v = \frac{1}{2}at^2 \), (c) \( t = \frac{v}{a} \), (d) \( v^2 = 2ax \).
5. Is it possible for two quantities to (a) have the same units but different dimensions or (b) have the same dimensions but different units? Explain.
6. Give an order-of-magnitude estimate for the time in seconds of the following:
   (a) a year; (b) a baseball game; (c) a heartbeat;
   (d) the age of Earth; (e) the age of a person.
7. Give an order-of-magnitude estimate for the length in meters of the following:
   (a) a person; (b) a fly; (c) a car; (d) a 747 airplane;
   (e) an interstate freeway stretching coast-to-coast.

Problems

Note: IP denotes an integrated conceptual/quantitative problem. BIO identifies problems of biological or medical interest.

Blue bullets (•, ••, •••) are used to indicate the level of difficulty of each problem.

Section 1-2 Units of Length, Mass, and Time

1. * The movie Spiderman broke all box-office records by bringing in more than $114,000,000 in its opening weekend. Express this amount in (a) gigadollars and (b) teradollars.
2. * A human hair has a thickness of about 70 \( \mu \)m. What is this in (a) meters and (b) kilometers?
3. * The speed of light is approximately 0.3 Gm/s. Express the speed of light in meters per second.
4. * A computer can do 2 gigacalculations per second. How many calculations can it do in a microsecond?

Section 1-3 Dimensional Analysis

5. * Velocity is related to acceleration and distance by the following expression, \( v^2 = 2ax \). Find the power \( p \) that makes this equation dimensionally consistent.
6. * Acceleration is related to distance and time by the following expression, \( a = 2x/t^2 \). Find the power \( p \) that makes this equation dimensionally consistent.
7. * Show that the equation \( v = v_0 + at \) is dimensionally consistent. Note that \( v \) and \( v_0 \) are velocities and that \( a \) is an acceleration.
8. * Newton’s second law (to be discussed in Chapter 5) states that acceleration is proportional to the force acting on an object and is inversely proportional to the object’s mass. What are the dimensions of force?
9. * The time \( T \) required for one complete oscillation of a mass \( m \) on a spring of force constant \( k \) is
   \[ T = 2\pi \sqrt{\frac{m}{k}} \]
   Find the dimensions \( k \) must have for this equation to be dimensionally correct.

Section 1-4 Significant Figures

10. * The first several digits of \( \pi \) are known to be \( \pi = 3.141592653 58979 \ldots \). What is \( \pi \) to (a) three significant figures, (b) five significant figures, and (c) seven significant figures?
11. * The speed of light to five significant figures is \( 2.9979 \times 10^8 \) m/s. What is the speed of light to three significant figures?
12. * A parking lot is 115.1 m long and 39.24 m wide. What is the perimeter of the lot?
13. * On a fishing trip you catch a 2.45-lb bass, a 10.1-lb rock cod, and a 16.13-lb salmon. What is the total weight of your catch?
14. * How many significant figures are there in (a) 0.000054, (b) 3.001 \times 10^5 ?
15. * What is the area of a circle of radius (a) 5.142 m and (b) 1.7 m?
Section 1-5 Converting Units

16. • The Eiffel Tower is 301 m high. What is its height in feet?
17. • (a) Calculate the volume of the warehouse in Example 1-2 in cubic feet. (b) Convert your result from part (a) to cubic meters.
18. • The Ark of the Covenant is described as a chest of acacia wood 2.5 cubits in length and 1.5 cubits in width and height. Given that a cubit is equivalent to 17.7 in., find the volume of the ark in cubic feet.
19. • How long does it take for radiation from a cesium-133 atom to complete 1 million cycles?
20. • Water going over Angel Falls in Venezuela, the world’s highest waterfall, drops through a distance of 3212 ft. What is this distance in km?
21. • An electronic advertising sign repeats a message every 8 seconds, day and night, for a week. How many times did the message appear on the sign?
22. • What is the conversion factor needed to convert seconds to years?
23. • The Star of Africa, a diamond in the royal scepter of the British crown jewels, has a mass of 530.2 carats, where 1 carat = 0.2 g. Given that 1 kg has an approximate weight of 2.21 lb, what is the weight of this diamond in pounds?
24. • IP Many highways have a speed limit of 55 mi/h. (a) Is this speed greater than, less than, or equal to 55 km/h? (b) Find the speed limit in km/h that corresponds to 55 mi/h.
25. • What is the speed in miles per hour of a beam of light traveling at 3.00 \times 10^8 m/s?
26. • Kangaroos have been clocked at speeds of 65 km/h. What is their speed in mi/h?
27. • ** IP Suppose 1.0 cubic meter of oil is spilled into the ocean. Find the area of the resulting slick, assuming that it is one molecule thick, and that each molecule occupies a cube 0.50 \mu m on a side.
28. • ** IP (a) A standard sheet of paper measures 8 1/2 by 11 inches. Find the area of each such sheet of paper in m². (b) A second sheet of paper is half as long and half as wide as the one described in part (a). By what factor is its area less than the area found in part (a)?
29. • ** BIO Nerve impulses in giant axons of the squid can travel with a speed of 20.0 m/s. How fast is this in (a) ft/s and (b) mi/h?
30. • ** The acceleration of gravity is approximately 9.81 m/s². (a) How much does the brain’s mass increase in one day? (b) How long does it take for the brain’s mass to increase by 0.0025 kg?

Section 1-6 Order-of-Magnitude Calculations

31. • Give a ballpark estimate of the number of seats in a typical major league ballpark.
32. • Milk is often sold by the gallon in plastic containers. Estimate the number of gallons of milk that are purchased in the United States each year. What approximate weight of plastic does this represent?
33. • NEW YORK is roughly 3000 miles from Seattle. When it is 1600 A.M. in Seattle, it is 1:00 P.M. in New York. Using this information, estimate (a) the rotational speed of the surface of Earth, (b) the circumference of Earth, and (c) the radius of Earth.
34. • You’ve just won the $1 million cash lottery, and you go to pick up the prize. What is the approximate weight of the cash if you request payment in (a) quarters or (b) dollar bills?

General Problems

35. • ** A Porsche can accelerate at 12 m/s². What is this in (a) ft/s² and (b) km/h²?
36. • ** BIO Type A nerve fibers in humans can conduct nerve impulses at speeds up to 140 m/s. (a) How fast are the nerve impulses in miles per hour? (b) How far (in meters) can the impulses travel in 5.0 ms?

The impulses in these nerve axons, which carry commands to the skeletal muscle fibers in the background, travel at speeds of up to 140 m/s. (Problem 36)